The crystal was mounted on a glass fibre. Data were collected at a scan speed of  $16.0^{\circ}$  min<sup>-1</sup> in  $\omega$ ;  $\omega$  scans of several intense reflections, made prior to data collection, had an average width at half-height of  $0.33^{\circ}$  with a take-off angle of  $6.0^{\circ}$ . The weak reflections  $[I < 15\sigma(I)]$  were rescanned to a maximum of five scans and the counts were accumulated to ensure good counting statistics. Stationary background counts were recorded on each side of the reflection. The ratio of peak to background counting time was 2:1.

The structure was solved by direct methods (Fan, 1991) and expanded using Fourier techniques (Beurskens *et al.*, 1992). The non-H atoms were refined anisotropically.

Anomalous dispersion effects were included in  $F_c$  (Ibers & Hamilton, 1964). The values for  $\Delta f'$  and  $\Delta f''$  were those of Creagh & McAuley (1992). The values for the mass attenuation coefficients were those of Creagh & Hubbell (1992). All calculations were performed using *TEXSAN* (Molecular Structure Corporation, 1985, 1992).

This work was supported by the NNSFC. The authors are grateful to Dr J. Sun at the Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, Shanghai 200032, China, for helpful discussions.

Lists of structure factors, anisotropic displacement parameters, Hatom coordinates and complete geometry have been deposited with the IUCr (Reference: BM1085). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

#### References

Beurskens, P. T., Admiraal, G., Beurskens, G., Bosman, W. P., Garcia-Granda, S., Gould, R. O., Smits, J. M. M. & Smykalla, C. (1992).
The DIRDIF Program System. Technical Report. Crystallography Laboratory, University of Nijmegen, The Netherlands.

Chuit, C., Corriu, R. J. P., Reye, C. & Young, J. C. (1993). Chem. Rev. 93, 1371–1448.

Creagh, D. C. & Hubbell, J. H. (1992). International Tables for Crystallography, Vol. C, pp. 200–206. Dordrecht: Kluwer Academic Publishers.

Creagh, D. C. & McAuley, W. J. (1992). International Tables for Crystallography, Vol. C, pp. 219–222. Dordrecht: Kluwer Academic Publishers.

Fan, H. F. (1991). Structure Analysis Programs with Intelligent Control. Rigaku Corporation, Tokyo, Japan.

Frye, C. L. (1970). J. Am. Chem. Soc. 92, 1205-1210.

Holmes, R. R. (1990). Chem. Rev. 90, 7-31.

Holmes, R. R., Day, R. O., Harland, J. J., Sau, A. C. & Holmes, J. M. (1984). Organometallics, 3, 341-353.

Holmes, R. R. & Deiters, J. A. (1977). J. Am. Chem. Soc. 99, 3318-3326

Hosomi, A., Kohra, S., Ogata, K., Yauagi, T. & Tominaga, Y. (1990).
J. Org. Chem. 55, 2415–2420.

Ibers, J. A. & Hamilton, W. C. (1964). Acta Cryst. 17, 781.

Johnson, C. K. (1965). ORTEP. Report ORNL-3794, Oak Ridge national Laboratory, Tennessee, USA.

Molecular Structure Corporation (1985, 1992). TEXSAN. TEXRAY Structure Analysis Package. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.

Small, J. H., McCord, D. J., Greaces, J. & Shea, K. J. (1995). J. Am. Chem. Soc. 117, 11588–11589.

Tacke, R., Becht, J., Lopez-Mras, A. & Sperlich, J. (1993). J. Organomet. Chem. 446, 1–8.

Tacke, R., Lopez-Mras, A. & Jones, P. G. (1994). Organometallics, 13, 1617–1623.

Tacke, R., Mühleisen, M. & Jones, P. G. (1994). Angew. Chem. Int. Ed. Engl. 33, 1186–1188.

Tamao, K., Nagata, K., Asahara, M., Kawachi, A., Ito, Y. & Shiro, M. (1995). J. Am. Chem. Soc. 117, 11592-11593.

Acta Cryst. (1996). C52, 3181-3183

## Oxopregnenyl 4-Allyloxybenzoate

DAVID W. TOMLIN,<sup>a</sup> TIMOTHY J. BUNNING,<sup>b\*</sup> COLIN M. MCHUGH<sup>c</sup> AND W. WADE ADAMS<sup>c</sup>

<sup>a</sup>Technical Management Concepts Inc., PO Box 340345, Beavercreek, OH 45434, USA, <sup>b</sup>Science Applications International Corporation, 101 Woodman Drive, Suite 103, Dayton, OH 45431, USA, and <sup>c</sup>Wright Laboratory, Materials Directorate, Wright-Patterson AFB, OH 45433, USA. E-mail: bunnintj@ml.wpafb.af.mil

(Received 11 March 1996; accepted 30 April 1996)

#### **Abstract**

The molecule of 20-oxo-5-pregnen-3-yl 4-allyloxy-benzoate,  $C_{31}H_{40}O_4$ , is extended, with the benzoate moiety twisted by  $34.0\,(1)^\circ$  from the least-squares plane through the tetracyclic core. The extended molecules pack in a parallel manner with respect to one another.

#### **Comment**

Polysiloxane backbones with pendant cholesterol ester groups are well known to produce liquid-crystalline behavior (Shibaev & Freidzon, 1989). The phase behavior of macromolecules based on liquid-crystalline cyclosiloxane backbones and cholesterol-based mesogens has also been examined extensively (Bunning & Kreuzer, 1995; Kreuzer et al., 1991; Gresham et al., 1994). These structures can be quenched into glassy transparent films which show promise for filter and storage applications. X-ray analysis of these materials has shown that a complex molecular architecture is present wherein two types of layered packing domains occur, each exhibiting characteristic X-ray diffraction features (Bunning, Klei, Samulski, Crane & Linville, 1991; Bunning, Klei, Samulski, Adams & Crane, 1993; Bunning, Socci, Farmer, Campbell & Adams, 1996). In systems with long spacer groups (the unit used to link the backbone to the tetracyclic core of the lathshaped cholesterol molecule), a reflection corresponding approximately to the length of the extended mesogen is observed. In short spacer group compounds, a second reflection indicative of partial interdigitation of mesogens is also observed. Both reflections, corresponding to  $C_{31}H_{40}O_4$ 

fully and partially interdigitated mesogens, appear with intermediate length spacer groups.

Previous examinations of two cholesteryl esters (Socci et al., 1995; Chabinyc et al., 1995) indicate some correlation between the solid-state conformation and the structure in the liquid-crystalline phase. In these two compounds, the length of the spacer unit was varied. In our continuing efforts to understand the key molecular interactions, several new steroidal mesogens, with different length and functionality units pendant on the  $17\beta$ -C atom of the tetracyclic core, were synthesized. The structure of an allyloxybenzoate-substituted pregnenyl compound, (I), in which the  $C_8H_{17}$  tail of cholesterol is replaced by a  $C(=O)CH_3$  tail, is examined here.

Bond lengths and angles in (I) are normal. The conformation of the tetracyclic core is consistent with that found in other cholesterol analogs (Guerina & Craven, 1979; Sawzik & Craven, 1980; Chabinyc et al., 1995; Socci et al., 1995). The benzoate moiety is twisted out of the least-squares plane through the tetracyclic core by 34.0 (1)°. The molecules are packed in a parallel manner (Fig. 2). The long axis of the mol-

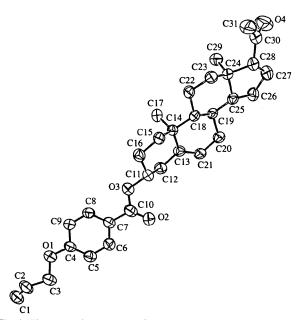


Fig. 1. The molecular structure of the title compound with the atomnumbering scheme. Displacement ellipsoids are drawn at the 50% probability level. H atoms have been omitted for clarity.

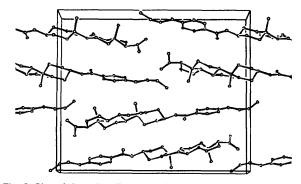


Fig. 2. Plot of the unit cell; the a axis is nearly perpendicular to the plane of the paper and the b and c axes horizontal and vertical, respectively.

ecule makes an angle of  $11.0^{\circ}$  with the (001) plane. The completely interdigitated packing corresponds well to the layer spacing observed in the liquid-crystalline phase (24.7 Å).

## **Experimental**

The title compound was synthesized *via* a room-temperature esterification reaction according to McHugh & Bunning (1995).

### Crystal data

$C_{31}H_{40}O_4$ $M_r = 476.7$ Orthorhombic $P2_12_12_1$ a = 7.460 (1) Å b = 17.118 (3) Å c = 20.707 (3) Å $V = 2644.3 (7) Å^3$ Z = 4 $D_x = 1.197 \text{ Mg m}^{-3}$ $D_m = 1.20 \text{ Mg m}^{-3}$ $D_m \text{ measured by flotation in}$	Mo $K\alpha$ radiation $\lambda = 0.71073 \text{ Å}$ Cell parameters from 25 reflections $\theta = 4-11^{\circ}$ $\mu = 0.08 \text{ mm}^{-1}$ T = 296  K Parallelepiped $0.45 \times 0.40 \times 0.30 \text{ mm}$ Colorless
$D_m$ measured by flotation in a 45wt% sucrose solution	

#### Data collection

Enraf-Nonius CAD-4	2701 observed reflections
diffractometer	$[I > 3\sigma(I)]$
$\omega/2\theta$ scans	$R_{\rm int}=0.074$
Absorption correction:	$\theta_{\text{max}} = 29.97^{\circ}$
Gaussian	$h = 0 \rightarrow 10$
$T_{\min} = 0.9688, T_{\max} =$	$k = -24 \rightarrow 24$
0.9736	$l = -29 \rightarrow 29$
16 410 measured reflections	2 standard reflections
4320 independent reflections	frequency: 240 min
	intensity decay: 0.9%

## Refinement

$(\Delta/\sigma)_{\rm max} = 0.002$
$\Delta \rho_{\rm max} = 0.421  {\rm e  \AA^{-3}}$
$\Delta \rho_{\min} = -0.447 \text{ e Å}^{-3}$
Extinction correction: none

2701 reflections 316 parameters H-atom positions calculated and refined as riding  $w = 1/[\sigma^2(F) + 0.000025F^2]$  Atomic scattering factors from *International Tables* for X-ray Crystallography (1974, Vol. IV, Tables 2.2B and 2.3.1)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (Å<sup>2</sup>)

 $U_{\text{eq}} = (1/3) \sum_{i} \sum_{j} U_{ij} a_i^* a_i^* \mathbf{a}_i . \mathbf{a}_j.$  $U_{\text{eq}}$  0.069 (2) 0.5654 (2) 0.6617 (4) -0.4059(2)01 0.061(2) O3 0.4420(3)0.6217(2) -0.1197(1)O2 0.1801 (4) 0.6560(2)-0.1645(2)0.084(2)0.4593 (2) 0.2322 (5) 0.107(3)04 0.6661 (3) CI -0.5679(3)0.081(3)0.6926 (7) 0.5277 (3) C2 0.7137 (6) 0.5706(3) -0.5185(2)0.073(3)C3 C4 C5 -0.4669(2)0.065(3)0.5764 (6) 0.5802(3)0.5796 (2) 0.051(2)0.5701 (5) -0.3511(2)-0.3480(2)0.3932 (5) 0.6019(3) 0.059(2)C6 0.3194 (5) 0.6176(3)-0.2879(2)0.057(3)**C**7 0.4153 (5) -0.2319(2)0.051(2) 0.6109(2)C8 -0.2368(2)0.5919 (5) 0.5864(2)0.056(2)C9 0.058(3)0.6688 (5) 0.5716(3) -0.2937(2)C10 0.3305 (6) 0.6314(3) -0.1703(2)0.059(3)C11 0.3863 (5) 0.6495 (2) -0.0563(2)0.052(2)0.2536 (5) 0.057(2)C12 0.5943 (2) -0.0249(2)0.045(2) C13 0.2179(5)0.6156(2) 0.0446 (2) C14 0.3822 (4) 0.6277 (2) 0.0868 (2) 0.043(2) C15 0.5125 (4) 0.6820(2)0.0508(2) 0.050(2)C16 0.5555 (5) 0.6568 (2) -0.0181 (2) 0.058(2)0.0981(2) 0.057(2) 0.5482(2)C17 0.4725 (5) C18 0.3308 (4) 0.6666 (2) 0.1513(2) 0.042(2)C19 0.1605 (4) 0.6318(2) 0.1815 (2) 0.042(2) 0.0053 (4) 0.1339(2)0.050(2)0.6397(2)C20 0.050(2) 0.0670(2) C21 0.0536 (4) 0.6237 (2) C22 0.4847 (4) 0.6705 (3) 0.1999(2)0.056(2)C23 0.4353 (5) 0.7085(2)0.2637 (2) 0.056(2)0.2760 (5) 0.2946(2)0.044(2)0.6683(2)C24 0.043(2) C25 0.1222 (4) 0.6704(2) 0.2453(2)C26 -0.0423(5)0.6470(2)0.2848(2)0.060(3)C27 -0.0042(5)0.6782(3) 0.3527(2)0.065(3) 0.052(2) 0.1847 (5) 0.7113(2)0.3519(2)C28 0.3149(2) 0.066(3) C29 0.3234 (6) 0.5847(3)C30 0.2811 (6) 0.7060(3)0.4148(2)0.068(3)C31 0.4454 (7) 0.7559(3)0.4227(3)0.094(4)

Table 2. Selected torsion angles (°)

C3-O1-C4-C5	5.5 (6)	C22—C18—C19—C20	172.3 (3)
C11—O3—C10—O2	-6.6(6)	C22—C23—C24—C29	66.5 (4)
C6C7C10O2	1.9 (7)	C23—C24—C25—C26	-167.1 (3)
C12—C13—C14—C17	-70.2(4)	C27—C28—C30—O4	14.0 (7)
C21—C13—C14—C15	-131.2(4)	C27—C28—C30—C31	-163.9(4)

All non-H atoms were refined anisotropically by full-matrix least-squares methods.

Data collection: CAD-4 Software (Enraf-Nonius, 1989). Cell refinement: CAD-4 Software. Data reduction: Xtal3.2 DIFDAT ADDREF ABSORB SORTRF (Hall, Flack & Stewart, 1992). Program(s) used to solve structure: Xtal3.2 GENTAN. Program(s) used to refine structure: Xtal3.2 CRYLSQ. Molecular graphics: Xtal3.2 ORTEP. Software used to prepare material for publication: Xtal3.2 BONDLA CIFIO.

Lists of structure factors, anisotropic displacement parameters, Hatom coordinates and complete geometry have been deposited with the IUCr (Reference: HA1161). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

#### References

Bunning, T. J., Klei, H. E., Samulski, E. T., Adams, W. W. & Crane, R. L. (1993). *Mol. Cryst. Liq. Cryst.* 231, 163–174. Bunning, T. J., Klei, H. E., Samulski, E. T., Crane, R. L. & Linville,

R. J. (1991). Liq. Cryst. 10, 445-446. Bunning, T. J. & Kreuzer, F. H. (1995). TRIP, 3, 318-323.

Bunning, T. J., Socci, E. P., Farmer, B. L., Campbell, A. L. & Adams, W. W. (1996). *Mol. Cryst. Liq. Cryst.* In the press.

Chabinyc, M. L., Fratini, A. V., Socci, E. P., Farmer, B. L., Bunning, T. J. & Adams, W. W. (1995). Acta Cryst, C51, 1444-1447.

Enraf-Nonius (1989). CAD-4 Software. Version 5. Enraf-Nonius, Delft, The Netherlands.

Guerina, N. G. & Craven, B. M. (1979). J. Chem. Soc. Perkin Trans. 2, pp. 1414–1419

Hall, S. R., Flack, H. D. & Stewart, J. M. (1992). Editors. Xtal3.2 Reference Manual. Universities of Western Australia, Australia, Geneva, Switzerland, and Maryland, USA.

McHugh, C. M. & Bunning, T. J. (1995). Am. Chem. Soc. Polym. Prep. 36, 379–380.

Shibaev, V. P. & Freidzon, Ya. S. (1989). Side-Chain Liquid Crystal Polymers, edited by C. B. McArdle, pp. 260-286. Glasgow: Blackie.
Socci, E. P., Farmer, B. L., Chabinyc, M. L., Fratini, A. V., Bunning, T. J. & Adams, W. W. (1995). Acta Cryst. C51, 888-891.

Sawzik, P. & Craven, B. M. (1980). Acta Cryst. B36, 215-218.

Acta Cryst. (1996). C52, 3183-3185

# A Bicyclo[2.2.1]heptenone System

KRAIG A. WHEELER AND CHRISTOPHER J. CARROW

Department of Chemistry, Delaware State University, Dover, DE 19901, USA. E-mail: kwheeler@desu.edu

(Received 4 December 1995; accepted 22 July 1996)

#### **Abstract**

The crystal and molecular structure of the title compound, 1,2,3,4-tetraphenyl-5-(4-pyridyl)bicyclo[2.2.1]-hept-2-en-7-one,  $C_{36}H_{27}NO$ , has been determined to confirm the stereochemistry and investigate its potential inclusion behavior.

#### Comment

Interest in the systematic design and application of organic clathrates has increased markedly in recent years. Studies have shown that sterically encumbered compounds may exhibit inclusion behavior without the cohesive force of hydrogen bonding (Atwood, Davies & MacNicol, 1984, 1991). These inclusion complexes, stabilized by weak van der Waals interactions, are controlled primarily by the topological complementarity of the guest and host molecules.

1,2,3,4-Tetraphenyl-5-(4-pyridyl)bicyclo[2.2.1]hept-2-en-7-one, (I), was prepared by a Diels-Alder reaction (Wasserman, 1965; Huisgen, Grashey & Sauer,